

> Deduction of Wein's law from Planck's law:-

From Planck's radiation law is

$$E_{\lambda} d\lambda = \frac{8\pi hc}{\lambda^5} \frac{d\lambda}{(e^{hc/\lambda kT} - 1)} \quad \text{①} \quad \left[\begin{array}{l} \text{in range } \lambda \text{ and} \\ \lambda + d\lambda \end{array} \right]$$

$$E_{\nu} d\nu = \frac{8\pi h \nu^3}{c^3} \cdot \frac{1}{(e^{h\nu/kT} - 1)} d\nu \quad \left[\begin{array}{l} \text{in range } \nu \text{ and} \\ \nu + d\nu \end{array} \right] \quad \text{②}$$

For small temperature, λT is small and for shorter wavelength $e^{hc/\lambda kT}$ becomes large compared to unity and Planck's law reduces to

$$E_{\lambda} d\lambda = \frac{8\pi hc}{\lambda^5} \frac{d\lambda}{e^{hc/\lambda kT}}$$

$$E_{\lambda} d\lambda = \frac{8\pi hc}{\lambda^5} e^{-hc/\lambda kT} d\lambda$$

which is Wein's law of radiation for shorter wavelength.

> Deduction of Rayleigh - Jeans law from Planck's law

From Planck's radiation law

$$E_{\lambda} d\lambda = \frac{8\pi hc}{\lambda^5} \frac{d\lambda}{(e^{hc/\lambda kT} - 1)}$$

For large temperature ΔT is large and also for longer wavelength $e^{hc/\lambda KT}$ can be approximated as $(1 + \frac{hc}{\lambda KT})$ ^②

$$\text{i.e. } e^{hc/\lambda KT} = 1 + \frac{hc}{\lambda KT} + \frac{(\frac{hc}{\lambda KT})^2}{2!} + \dots = 1 + \frac{hc}{\lambda KT}$$

Hence Planck's law reduces to

$$E_{\lambda} d\lambda = \frac{8\pi hc}{\lambda^5} \cdot \frac{d\lambda}{\left(1 + \frac{hc}{\lambda KT} - 1\right)}$$

$$= \frac{8\pi hc}{\lambda^5} \cdot \frac{\lambda KT}{hc} \cdot d\lambda$$

$$E_{\lambda} d\lambda = \frac{8\pi KT}{\lambda^4} d\lambda$$

This is the Rayleigh-Jeans law.

